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States of Matter; 2 3.1 Solids,

Liquids Gases. Objectives ; 1.

Explain how shape and volume can be used to classify materials ; 2.

Describe how kinetic theory and forces of attraction can be used to explain ...

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4 Atomic Structure Atom V Element

ATOMIC THEORY Democritus (4th

century B.C) said the universe was

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made of invisible units called atoms
In 1808, an Englishman named John Dalton proposed a theory: 1. every element is made of tiny unique particles called atoms that cannot be subdivided 2. atoms of the same element have the same ... Pearson

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Help ... Freezing point is a physical prop- erty and ability to combine
with oxygen is a chemical property.
The melting of ice to water is an
example of a physical change, and
the burn- ing of wood is an example
of a chemical change. A mixture is
made of two or more substances. A
substance is made of only one type
of matter. Chapter 1 Introduction to
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Assessing the accuracy of species distribution models: Prevalence, kappa and the true skill statistic (TSS) . Effective Key Parameter Determination for an Automatic

... Which of the following is not directly part of the physical environment? An understanding of Earth is essential for ... All science is based on the assumption that the natural world behaves in a consistent and predictable manner that is comprehensible through

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